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Haloarenes are less
polar than

haloalkanes and are
insoluble in water.

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This is because of lack of hydrogen bonding. As a result, the attractive forces in haloarenes—water system remain less than the attractive forces in H₂O molecules which are hydrogen bonded.

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10.1: Name the following halides according to IUPAC system and classify them as alkyl, allyl, benzyl (primary, secondary, tertiary), vinyl or aryl halides :

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bond (carbon-halogen
bond) of alkyl halide is

polarised. Halogen

atom carries a partial

negative charge

Carbon atom carries a

partial positive

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Haloalkanes are

classified as fluoro,

chloro, bromo or iodo

compounds according

to the type of halogen

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present and as
mono-, di- tri-, tetra-
haloalkanes, etc.,
according to the one,
two, three, four, etc.,
halogen atoms
respectively present
in their molecule. 2.

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Haloalkanes and Haloarenes. The replacement of hydrogen atom(s) in hydrocarbon, aliphatic or aromatic, by halogen atom(s) results in the formation of alkyl halide (haloalkane) and aryl halide (haloarene), respectively.

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Classification of Halogen Derivatives. And Haloarenes

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Haloalkanes and ...
Haloalkanes are the
halogen derivatives of
hydrocarbons and
these are of the
following types
depending upon the
number of hydrogen
atoms present in

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them. Haloarenes are aromatic compounds in which the halogen atom is attached directly to the carbon atom of the aromatic ring. The solutions of chapter 10 also give exemplary problems, HOTS (Higher Order Thinking Skills), MCQs exercises, assignments and worksheets which will

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help you to learn the
concept
comprehensively and
to make haloalkanes
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replacement of

hydrogen atom (s) in

a hydrocarbon,

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Aliphatic or aromatic, by halogen atom (s), results in the formation of alkyl halide. (haloalkane) and aryl halide (haloarene), respectively.

Haloalkanes contain halogen atom (s) attached to the sp^3 hybridized carbon atom of an alkyl group whereas haloarenes

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contain halogen atom (s) attached to sp^2 hybridized carbon atom (s) of an aryl group.

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C-X bond in alkyl halides: X is more electronegative than carbon. So, the C-X bond is polarized with C having a partial positive charge and X having a partial negative charge.

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Preparation of
Haloalkanes II: 3:

Properties of
Haloalkanes I: 4:

Properties of
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Properties of
Haloalkanes III: 6:

Allyl Halides and
Dihalides I: 7:

Reactions Of

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Dihalides I: 8:

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Haloarenes I: 9:

Properties of

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Notes 1. Haloalkanes are classified as fluoro, chloro, bromo or iodo compounds according to the type of halogen present and as mono-, di- tri-, tetra- haloalkanes, etc., according to the one, two, three, four, etc., halogen atoms

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respectively present
in their molecule. 2.

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