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  - B. Neutral solution
  - C. Saturated solution
  - D. Supersaturated solution.
10. Protein is a

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polymer formed  
from the linkage  
of A. Amino acid  
molecules. B.  
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molecules. C.  
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D.

Monosaccharides.  
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. SECTION A. 1.  
Identify the  
solid remaining



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NaBr and 0.20  
mole of MgBr<sub>2</sub>  
in 2.0 dm<sup>3</sup> of  
water is to be  
analysed. ...

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SECTION A. 1.

Identify the solid remaining when each of the following is heated. ...

State whether R would be expected to form acidic or basic oxide. (b) (i) State two assumptions of the ...

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An aqueous solution that contains 1 mol (342 g) of sucrose in enough water to give a final volume of 1.00 L has a sucrose concentration of



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1.00 mol/L or  
1.00 M. In  
chemical  
notation, square  
brackets around  
the name or  
formula of the  
solute represent  
the molar  
concentration of  
a solute.

Therefore,  
[sucrose] =  
1.00M.

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A buffer solution (more precisely, pH buffer or hydrogen ion buffer) is an aqueous solution consisting of a

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mixture of a weak acid and its conjugate base, or vice versa. Its pH changes very little when a small amount of strong acid or base is added to it. Buffer solutions are used as a means of keeping pH at

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a nearly  
constant value  
in a wide  
variety of  
chemical  
applications.

Buffer solution  
- Wikipedia

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required to  
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