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Chemistry** ~~Modern Chemistry Chapter
7-2~~

Modern Chemistry • CHAPTER 7
HOMEWORK 7-2 (pp. 206–211)
VOCABULARY Match the symbol with
the name that describes it. 1. Br⁻
_____ a. nitrate 2. OH⁻ _____ b.
chlorite 3. NO₂ - _____ c. hydroxide

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4. NO₃ – _____ d. bromide 5. O₂ –
_____ e. nitrite 6. CN – _____ f.
ammonium 7. ClO₂ – _____ g. oxide
8. NH₄ + _____ h. cyanide GRAPHIC
ORGANIZER

~~HOMEWORK 7-2~~

Ba(OH)₂ —The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for every atom of barium. (NH₄)₂SO₄

HOMEWORK 7-2 Modern Chemistry
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3. 4 terms. Modern Chemistry Chapter
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challenging the brain to think better and faster can be undergone by some ways. Experiencing, listening to the extra experience, adventuring, studying, training, and more practical deeds may support you to improve.

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this set (32) monatomic ion. an ion
formed from a single atom. (remember
to change nonmetals to -ide ending)

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1) Calculate the amount of each element in sample. 2) Convert the mass composition of each element in moles. %/mass divide smallest number. If given the % of each element in a compound, be able to calculate the empirical formula of the compound. %/ element mass; divide

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answers by the smallest number.

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$2 = 0.947 \text{ atm}$; $P_2 \text{ of } O_2 = 0.987 \text{ atm}$

Unknown: $V_2 \text{ of } O_2 \text{ in mL}$ VPLAN

Rea r ange the equ tion for Boyl 's l w

($P_1 V_1 = P_2 V_2$) to obtain V_2 . $V_2 = P_1 V_1 / P_2$

SSOLVE substitute values for

P_1 , V_1 , and P_2 to obtain the new

volume, V_2 . $V_2 = 2 (0.947$

$\text{atm})(150.0 \text{ mL } O_2) / 0.987 \text{ atm} =$

$144 \text{ mL } O_2$ CHECK YOUR WORK

When the p ress ue is inc ea sd ...

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~~Chemistry~~

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Chemistry Chapter 7Chemical
Formulas Chemical Compounds . A
chemical formula indicates the kind
and relative number of atoms in a
chemical compound. C₈H₁₈ (octane)
has 8 carbon and 18 hydrogen atoms.
2 Forming Ionic Compounds.
Compounds that have the elements
held together by

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