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carbon in $\text{CH}_3\text{C}\equiv\text{N}$ and the bond angle CCN are a. sp^2 , 180° . b. sp , 180° . c. sp^2 , 120° . d. sp^3 , 109° . 2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds

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that are linear. c. It has
two p orbitals. d.

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Trimethylamine is a pyramidal molecule. What is the hybridization at nitrogen? Question2: How many stereoisomers of phenylpropylene oxide are there? Question3: Which one of the Following is not

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2)Benzene 3)Naphthalene
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Solution: Dipole moment of
CCl₄ is 0. The order of
dipole moment is CCl₄ < CH

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3 $\text{Cl} < \text{CH}_2 < \text{Cl}_2 < \text{CHCl}_3$.

Hence option (3) is the answer. 4. Which one of the following compounds will not be soluble in sodium bicarbonate ? (1) Benzene sulphonic acid (2) Benzoic acid (3) o-Nitrophenol (4)

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compound is given below:
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Solutions methane, ethyne, and ethene.

From the list above, name a compound : (i) Formed by the dehydration of ethanol by concentrated sulphuric acid.

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