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~~*to 29*~~ **Solution Stoichiometry - Finding**

**Molarity, Mass & Volume**

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~~Stoichiometry Worksheets with Answer Keys~~  
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Balancing Equations and Simple Stoichiometry-  
KEY Balance the following equations: 1)  $1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$  2)  $2 \text{ C}_6\text{H}_{10} + 17 \text{ O}_2 \rightarrow 12 \text{ CO}_2 + 10 \text{ H}_2\text{O}$  3)  $1 \text{ HBr} + 1 \text{ KHCO}_3 \rightarrow 1 \text{ H}_2\text{O} + 1 \text{ KBr} + 1 \text{ CO}_2$  4)  $2 \text{ GaBr}_3 + 3 \text{ Na}_2\text{SO}_3 \rightarrow 1 \text{ Ga}_2(\text{SO}_3)_3 + 6 \text{ NaBr}$  5)  $3 \text{ SnO} + 2 \text{ NF}_3 \rightarrow 3 \text{ SnF}_2 + 1 \text{ N}_2\text{O}_3$  Using the following equation:  $2 \text{ NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{ H}_2\text{O} + \text{Na}_2\text{SO}_4$

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Mole Conversions and Stoichiometry Review

Worksheet. 1) Using the following equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + 2 \text{H}_2\text{O}$

How many grams of sodium sulfate will be formed

if you start with 200 grams of sodium

hydroxide and you have an excess of sulfuric

acid ( $\text{H}_2\text{SO}_4$ )? 2) Using the following

equation:  $\text{Pb}(\text{SO}_4)_2 + 4 \text{LiNO}_3 \rightarrow \text{Pb}(\text{NO}_3)_4 +$

$2 \text{Li}_2\text{SO}_4$ . How many grams of lithium

nitrate will ...

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Answers Balancing Equations and Simple  
Stoichiometry-KEY Balance the following  
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following equation:  $2 \text{ NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{ H}_2\text{O} + \text{Na}_2\text{SO}_4$

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So, you've finally, done it: You've entered  
the realm of stoichiometry. Or as some people

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pronounce it, "stoi-shee-oh-met-tree." Don't pronounce it that way, it'll make you sound silly. The actual pronunciation: "stoy-key-ah-meh-tree." Now that we've got that out of the way, let's learn about the magical world of stoichiometry! The magical world of stoichiometry In...

~~The magic of stoichiometry | The Cavalcade of Chemistry~~

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